



Pergamon

Tetrahedron Letters 41 (2000) 1165–1169

TETRAHEDRON
LETTERS

Solid-phase synthesis of 3-aminohydantoin, dihydrouracil, thiohydantoin and dihydrothiouracil derivatives

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Received 20 October 1999; accepted 6 December 1999

Abstract

An efficient and convenient route to 3-aminohydantoins and dihydrouracils using solid-phase chemistry has been developed. This methodology has the advantage of generality and diversity over solution-phase chemistry for the preparation of 3-aminohydantoins and dihydrouracils. © 2000 Elsevier Science Ltd. All rights reserved.

Hydantoin and aminohydantoin-containing heterocycles show a broad range of biological activity including anticonvulsant, antibacterial, antifungal, antiarrhythmic, antiherbicide, diuretic, metalloprotease inhibitory and central nervous system activity.¹ Recently, there has been growing interest in the development of solid-phase synthetic approaches to hydantoin derivatives, particularly those substituted at the *N*-1, *N*-3 and *C*-5 positions.^{2–10} In contrast to the many examples of the synthesis of hydantoin derivatives on solid-supports, there are only a few reports on the synthesis of 1-aminohydantoins¹¹ and 3-aminohydantoins.^{12,13} To the best of our knowledge, there are no reports on the solid-phase synthesis of 3-aminodihydrouracil and thiodihydrouracil derivatives. In the preceding paper, we reported a one-pot solution-phase synthetic route to 3-aminohydantoin and dihydrouracil derivatives. We describe herein the general solid-phase synthesis of 3-amino-hydantoins, -thiohydantoins, -dihydrouracils and -thiodihydrouracils.

Although our one-pot solution-phase methodology provides a simple route to 3-aminohydantoin and 3-aminodihydrouracil derivatives **I** (Fig. 1), there are some drawbacks related to it: First, the substituent at *N*-1 derives from an α -amino acid derivative and only a limited number of *N*-substituted α -amino acid derivatives is commercially available which limits the potential scope of this method. Secondly, the preparation and purification of compounds with basic substituents such as the *N*-alkylamine substituted 3-aminohydantoin derivatives **II** (Fig. 1) (and their starting materials, namely *N*-alkylamine substituted α -amino acids), is very tedious. In view of this, we deemed it of interest to extend our solution-phase chemistry to a solid-phase method for the preparation of 3-aminohydantoins and dihydrouracils. The synthesis is shown in Scheme 1.

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Table 1
Solid-phase synthesis of 3-aminohydantoin and dihydrouracil derivatives

Entry	Products ^a	Purity ^b	Yield ^c	Entry	Products ^a	Purity ^b	Yield ^c		
1		10a	97%	49%	13		10m	90%	58%
2		10b	95%	54%	14		10n	81%	54%
3		10c	90%	57%	15		10o	74%	45%
4		10d	93%	63%	16		10p	86%	57%
5		10e	96%	61%	17		10q	66%	21%
6		10f	86%	50%	18		10r	90%	32%
7		10g	95%	42%	19		10s	---	---
8		10h	97%	59%	20		11a	94%	46%
9		10i	98%	59%	21		11b	95%	52%
10		10j	79%	41%	22		11c	90%	22%
11		10k	83%	69%	23		11d	98%	42%
12		10l	86%	67%	24		11f	85%	45%

a. ¹H and ¹³C NMR, HRMS and mass spectral data were consistent with the indicated structures. *b.* Purity of products was assayed by HPLC (H₃PO₄/H₂O/CH₃CN: 0.1/94.9/5). *c.* Isolated yields were based on the loading of commercial resin **1**.

substantially (>90% by LC-MS and NMR) through a simple filtration and evaporation. Cyclization of 3-aminodihydrouracil derivatives **11** were also successful by heating in DMF at 90°C for 24 h.

Using the protocol¹⁴ described above, we synthesized a variety of 3-aminohydantoin and 3-aminodihydrouracil derivatives. The results are summarized in Table 1.

In most cases, the aromatic and heteroaromatic-substituted primary amines were smoothly coupled and underwent cyclization/cleavage to afford products **10a–10h** and **11a–11b** in good yield and high purity (entries 1–8, 20, 21). Similarly, incorporation of heterocyclic-substituted amines and alkyl amines were also successful (entries 9–16, 23). Using the same conditions, the 3-aminothiohydantoins and 3-aminodihydrothiouracils were obtained in lower yield (entries 18 and 22). Secondary amines provided products in 21 and 45% yields, respectively (entries 17 and 24). None of the desired product was obtained from a *tert*-alkyl-substituted amine (entry 19).

In summary, an efficient and convenient route to the synthesis of 3-aminohydantoins and dihydrouracils using solid-phase chemistry has been developed. This solid-phase method allows for easy purification of amine-containing compounds and provides for a diverse set of substituents at the 1-position which derive from readily available primary amines.

Acknowledgements

We are grateful to Dr. Jeff Ares and Professor Daniel M. Ketcha for their helpful discussions.

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- Experimental details for the synthesis of **11a**: (a) Preparation of Merrifield resin-bound acrylate ester **5**: To a solution of 1,3-diisopropyl carbodiimide (15 g, 119 mmol), acrylic acid (17 g, 208 mmol) and Merrifield resin (25 g, 20 mmol, loading level: 0.80 mmol/g) in methylene chloride (300 mL) was added DMAP (0.5 g, 4 mmol). The resulting mixture was shaken at room temperature for 24 h. Resin was collected on a glass filter and washed twice each with DMF, MeOH, DCM. The resin was dried to give the Merrifield resin-bound acrylate ester **5** (37 g, yield 94%). (b) Preparation of compound **11a**: The Merrifield resin-bound acrylate ester **5** (2 g, loading 0.8 mmol/g) was treated with DMSO (50 mL) and 2-(2-aminoethylamino)-5-nitropyridine (1.46 g, 8.0 mmol) and allowed to shake for 24 h at room temperature. Washing twice each with DMF, MeOH, DCM afforded resin **7** (where R=(5-nitro-2-pyridinyl)aminoethyl), which was then treated with Boc-hydrazinecarbonylimidazole (8 mmol) (prepared in situ as described in the preceding paper) in 40 mL of DMF at room

temperature for 24 h and washed twice each with DMF, MeOH, DCM afforded resin **9** (X=O). The resin **9** was placed in a flask with 40 mL of DMF and heated to 95°C for 24 h. After cooling, the resin was filtered, washed with small amounts of DMF, DCM, MeOH and the combined filtrates was concentrated. The residue was dissolved in 40 mL of EtOAc and filtered, concentrated in vacuo to give the desired product **11a** (289 mg, 46%, purity: >95% by LC-MS). MS m/z 395 (M+1); ^1H NMR (CDCl_3) δ 9.02 (d, 1H, $J=2.6$ Hz), 8.16, 8.14 (d \times d, 1H, $J_1=J_2=2.8$ Hz), 6.69 (s, 1H), 6.48 (d, 1H, $J=2.6$ Hz), 6.04 (br, 1H), 3.77 (m, 4H), 3.61 (m, 1H), 3.46 (m, 1H), 2.85 (m, 2H), 1.50 (s, 9H); ^{13}C NMR (CDCl_3) δ 167.5, 165.0, 154.7, 153.9, 144.5, 136.1, 110.0, 83.0, 48.6, 42.8, 41.2, 31.9, 28.4; HRMS calcd for $\text{C}_{16}\text{H}_{23}\text{N}_6\text{O}_6$ (M+H): 395.1679; found: 395.1668.